

BBA 41141

Similar effects on enzymic activity due to chemical modification of either of two sulphhydryl groups of myosin

It is well known that increasing concentrations of *p*-mercuribenzoate or *N*-ethylmaleimide¹ increase Ca^{2+} -activated ATPase activity and inhibit both the K^{+} -activated ATPase and the Ca^{2+} -activated ITPase activities of myosin. Higher concentrations of reagent inhibit all activity¹⁻³. These results have been interpreted in terms of two functionally different classes of sulphhydryl groups in myosin, the first (S_1) containing rapidly reacting groups and the second (S_2) containing slowly reacting groups. Two peptides, each containing single cysteine residues believed to correspond to S_1 and S_2 , have been isolated from myosin and their amino acid sequences determined⁴⁻⁷.

In previous work myosin has been chemically modified by the reaction of only S_1 or of both S_1 and S_2 , but no attempt has been made to specifically modify S_2 leaving S_1 unchanged. The present experiments have been designed to allow S_2 to react with *N*-ethylmaleimide but to leave S_1 unchanged. This has been achieved by the reaction of S_1 with a disulfide (5,5'-dithio-bis-(2-nitrobenzoate) (DTNB)), the reaction of S_2 with *N*-ethylmaleimide, and finally, regeneration of the thiol group at S_1 by reduction with 1,4-dithiothreitol. The experimental procedure and nomenclature used are illustrated in Table I.

TABLE I

METHOD OF MODIFICATION OF SH GROUPS OF MYOSIN

Reaction	SH groups modified
Myosin + DTNB \rightarrow DTNB- S_1 -myosin + 5-thio-2-nitrobenzoate	S_1
DTNB- S_1 -myosin + <i>N</i> -ethylmaleimide \rightarrow DTNB- S_1 , <i>N</i> -ethylmaleimide- S_2 -myosin	S_1 and S_2
DTNB- S_1 , <i>N</i> -ethylmaleimide- S_2 -myosin + 1,4-dithiothreitol \rightarrow <i>N</i> -ethylmaleimide- S_2 -myosin + 5-thio-2-nitrobenzoate + oxidized 1,4-dithiothreitol	S_2

The effects of each of these reactions on the ATPase and ITPase activities of myosin are shown in Table II. Modification with DTNB resulting in the formation of DTNB- S_1 -myosin produces 80-90% loss of K^{+} -activated ATPase and Ca^{2+} -activated ITPase, accompanied by a small (50% in 0.6 M KCl) increase in Ca^{2+} -ATPase activity. DTNB appears to be less effective than other sulphhydryl reagents in activating Ca^{2+} -ATPase of myosin; the largest activation observed with DTNB was 100% in contrast to a 500-1000% activation produced by *N*-ethylmaleimide. Modification of DTNB- S_1 -myosin with *N*-ethylmaleimide produces complete loss of enzymic activity. The addition of 1,4-dithiothreitol, which leads to the removal of 5-thio-2-nitrobenzoate from S_1 leaving *N*-ethylmaleimide- S_2 -myosin, restores Ca^{2+} -activated ATPase activity but not K^{+} -ATPase or Ca^{2+} -ITPase activities. Ca^{2+} -ATPase activity of *N*-ethylmaleimide- S_2 -myosin at low salt concentrations is 50% of the activity of native myosin, while at higher salt concentrations the activity is 260%

Abbreviation: DTNB, 5,5'-dithio-bis-(2-nitrobenzoate).

TABLE II

EFFECT OF SEQUENTIAL REACTION OF MYOSIN WITH DTNB, *N*-ETHYLMALEIMIDE AND 1,4-DITHIO-THREITOL ON ENZYMIC ACTIVITIES

Myosin (20 mg/ml) was treated with 1 mM DTNB in a solution containing 0.5 M KCl, 0.04 M Tris (pH 8.0). After 5 min at 0° the pH was reduced to 7.0 by the addition of 0.4 M imidazole (pH 6.0) and the myosin passed through a 1 cm × 10 cm column of Biogel P-60 which had been equilibrated with a solution containing 0.5 M KCl and 0.002 M *N*-tris-(hydroxymethyl)methyl-2-aminoethanesulfonic acid at pH 7.0. Myosin at this step is referred to as DTNB-S₁-myosin. DTNB-S₁-myosin (8.3 mg/ml) was treated with 2 mM *N*-ethylmaleimide in a solution containing 5 mM CaCl₂, 5 mM ADP, 0.35 M KCl and 0.02 M imidazole (pH 7.0) for 30 min at 0° and again passed through the Biogel column. Myosin at this step is referred to as DTNB-S₁, *N*-ethylmaleimide-S₂-myosin. 0.1 vol. of 0.1 M 1,4-dithiothreitol was added and the solution incubated for 30 min at 0°. Myosin at this step is referred to as *N*-ethylmaleimide-S₂-myosin. ATPase and ITPase activities were measured by determination of P_i released at 25° in 0.05 M Tris (pH 7.5). For K⁺-ATPase the assay systems also contained: 0.6 M KCl, 5 mM EDTA and 5 mM ATP; for Ca²⁺-ATPase, 10 mM CaCl₂ and 5 mM ATP; and for Ca²⁺-ITPase, 10 mM CaCl₂ and 5 mM ITP.

Myosin	Activity (μmoles P _i /mg myosin per min)			
	K ⁺ -ATPase		Ca ²⁺ -ATPase	
	0.6 M KCl	0.05 M KCl	0.6 M KCl	Ca ²⁺ -ITPase 0.05 M KCl
Native myosin	1.85	0.74	0.29	1.02
DTNB-S ₁ -myosin	0.09	0.94	0.44	0.13
DTNB-S ₁ , <i>N</i> -ethylmaleimide-S ₂ -myosin	0.00	0.02	0.09	0.00
<i>N</i> -Ethylmaleimide-S ₂ -myosin	0.27	0.38	0.75	0.06

of the activity of native myosin. The activation by KCl of the Ca²⁺-ATPase of *N*-ethylmaleimide-S₂-myosin is similar to that of *N*-ethylmaleimide-S₁-myosin⁸. In contrast the Ca²⁺-ATPase activity of native or DTNB-S₁-myosin is inhibited by increasing KCl concentrations^{9,10}. From these results it appears that reaction of either S₁ or S₂ with *N*-ethylmaleimide has essentially the same effect on enzymatic activity, viz. the loss of K⁺-ATPase and Ca²⁺-ITPase activities and the increase of Ca²⁺-ATPase activity at high KCl concentrations. The question could be raised whether the product of the treatment with DTNB, *N*-ethylmaleimide and 1,4-dithiothreitol, as described above has been correctly identified as *N*-ethylmaleimide-S₂-myosin. If DTNB-S₁-myosin underwent a disulfide exchange involving the translocation of 5-thio-2-nitrobenzoate from S₁ to S₂ before or during modification with *N*-ethylmaleimide, the procedure designed to produce *N*-ethylmaleimide-S₂-myosin might actually lead to the formation of *N*-ethylmaleimide-S₁-myosin. To confirm that our procedure does indeed produce a reaction of *N*-ethylmaleimide with S₂, the effect of ADP on the inactivation of DTNB-S₁-myosin and *N*-ethylmaleimide-S₂-myosin by further addition of *N*-ethylmaleimide was investigated, since in contrast to S₁, S₂ is known to react slowly with *N*-ethylmaleimide, and the reaction is accelerated by ADP¹¹. If S₁ were blocked and S₂ had not reacted, the reaction with *N*-ethylmaleimide should lead to a slow loss of Ca²⁺-ATPase activity which should be accelerated by ADP; on the other hand, if S₂ were blocked and S₁ free, the reaction with *N*-ethylmaleimide should produce a rapid loss of Ca²⁺-ATPase even in the absence of ADP. In fact, reaction of DTNB-S₁-myosin with *N*-ethylmaleimide, as indicated by the loss of Ca²⁺-ATPase activity, is slow and markedly accelerated by ADP (Fig. 1A), while the reaction of *N*-ethylmaleimide-S₂-myosin with *N*-ethylmaleimide is rapid in the ab-

sence of ADP (Fig. 1B). These facts strongly suggest that it is indeed S_2 which is modified by *N*-ethylmaleimide during the preparation of what we describe as *N*-ethylmaleimide- S_2 -myosin.

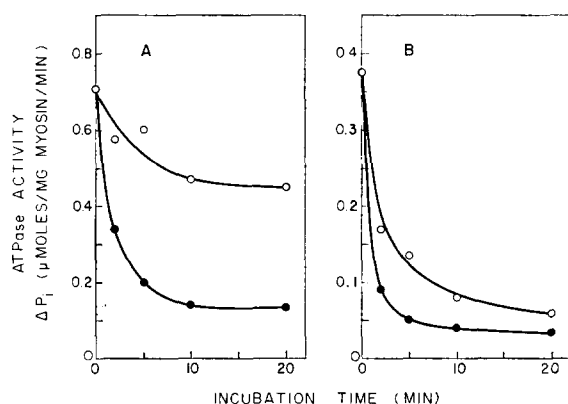


Fig. 1. Effect of Ca^{2+} and ADP on the rate of inactivation by *N*-ethylmaleimide of Ca^{2+} -activated ATPase activity of DTNB- S_1 -myosin and *N*-ethylmaleimide- S_2 -myosin. The preparation of DTNB- S_1 -myosin and *N*-ethylmaleimide- S_2 -myosin is described in the legend to Table II. The conditions for final reaction with *N*-ethylmaleimide were 0.39 M KCl, $1.5 \cdot 10^{-4}$ M *N*-ethylmaleimide, 0.015 M imidazole (pH 7.0) and, when added, 4 mM $CaCl_2$ and 4 mM ADP: (A) DTNB- S_1 -myosin (5 mg/ml); (B) *N*-ethylmaleimide- S_2 -myosin (3.2 mg/ml). ATPase activity was measured at 25° in a mixture containing 0.025 M KCl, 10 mM $CaCl_2$, 5 mM ATP, 0.05 M Tris (pH 7.5) and 0.1 mg myosin/ml. ○, reaction with *N*-ethylmaleimide in the absence of Ca^{2+} and ADP; ●, reaction in the presence of Ca^{2+} and ADP.

In the interpretation of these results, it is assumed that the sulfhydryl groups we refer to as S_1 and S_2 are the same as S_1 and S_2 referred to by other workers^{4,5,11,12}. Whether or not this assumption is correct, it is clear that there are two different classes of sulfhydryl groups of myosin having different reactivities toward sulfhydryl reagents, and chemical modification of either class has similar effects on enzymic activity. Modification of either S_1 or S_2 is sufficient to abolish K^+ activation of ATPase and Ca^{2+} activation of ITPase activity and to increase Ca^{2+} -ATPase activity. Modification of S_2 alone does not abolish Ca^{2+} -ATPase; thus this group, as such, is not essential for Ca^{2+} activation of ATPase activity. Activation of ATPase activity by Ca^{2+} requires the presence of one intact sulfhydryl group which may be either S_1 or S_2 .

The recent work of TROTTA *et al.*¹² suggests that S_1 and S_2 are not directly involved in the binding or hydrolysis of ATP by myosin since they apparently are not present in Subfragment 1 prepared by tryptic digestion of myosin. The sulfhydryl groups may be regarded as being present in a regulatory region whose conformation is changed by thiol reagents. Our findings that modification of either S_1 or S_2 produces similar effects on ATPase activity suggest that modification of either group alone alters activity by producing similar changes in the conformation of the regulatory region accompanied by a change in conformation at the active site such that K^+ activation of ATPase activity and Ca^{2+} activation of ITPase activity are lost, and Ca^{2+} activation of ATPase activity at high salt concentration is increased. In terms of this model modification of both sulfhydryl groups leads to a conformational change in the regulatory region that produces loss of all activity.

This work was supported by Grant H-5949 from the National Heart Institute, by grants from the Muscular Dystrophy Associations of America, Inc., the Life Insurance Medical Research Fund, the American Heart Association, the National Science Foundation, and by General Research Support Grant 1-S01-FR-05527 from the Division of Research Facilities and Resources, U.S. Public Health Service. This work was carried out during the tenure of an Established Investigatorship of the American Heart Association.

*Department of Muscle Research,
Retina Foundation, Institute of Biological and Medical Sciences;
Department of Neuropathology, Harvard Medical School,
Boston, Mass. 02114 (U.S.A.)*

J. C. SEIDEL

- 1 W. W. KIELLEY AND L. B. BRADLEY, *J. Biol. Chem.*, 218 (1956) 653.
- 2 J. J. BLUM, *Arch. Biochem. Biophys.*, 87 (1960) 104.
- 3 P. K. RAINFORD, K. HOTTA AND M. MORALES, *Biochemistry*, 3 (1964) 1213.
- 4 T. YAMASHITA, Y. SOMA, S. KOBAYASHI, T. SEKINE, K. TITANI AND K. NARITA, *J. Biochem. Tokyo*, 55 (1964) 576.
- 5 T. YAMASHITA, Y. SOMA, S. KOBAYASHI AND T. SEKINE, *J. Biochem. Tokyo*, 57 (1965) 460.
- 6 M. KIMURA AND W. W. KIELLEY, *Biochem. Z.*, 345 (1966) 188.
- 7 A. G. WEEDS AND B. S. HARTLEY, *Biochem. J.*, 107 (1968) 531.
- 8 T. SEKINE AND W. W. KIELLEY, *Biochim. Biophys. Acta*, 81 (1964) 336.
- 9 J. C. WARREN, L. STOWRING AND M. MORALES, *J. Biol. Chem.*, 241 (1966) 309.
- 10 J. C. SEIDEL, *J. Biol. Chem.*, 244 (1969) 1142.
- 11 T. SEKINE AND M. YAMAGUCHI, *J. Biochem. Tokyo*, 54 (1963) 196.
- 12 P. P. TROTTA, P. DREIZEN AND A. STRACHER, *Proc. Natl. Acad. Sci. U.S.*, 61 (1968) 659.

Received February 7th, 1969

Biochim. Biophys. Acta, 180 (1969) 216-219